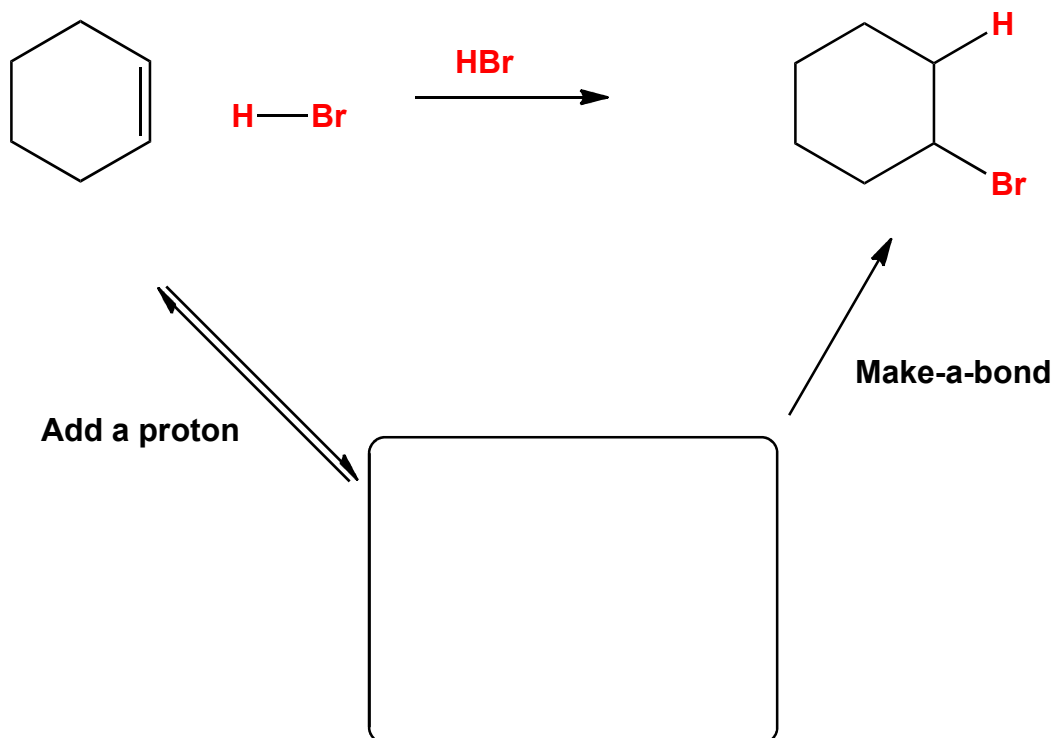
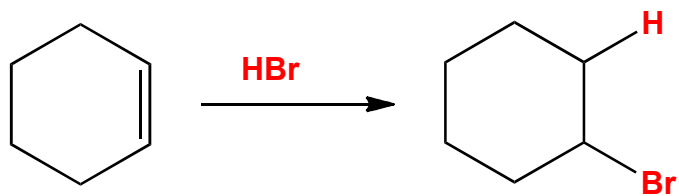
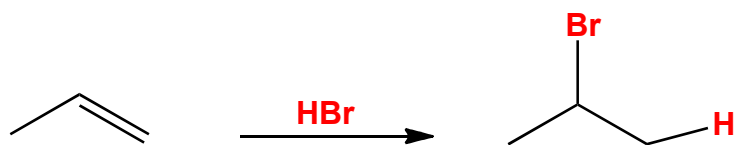


Reaction:

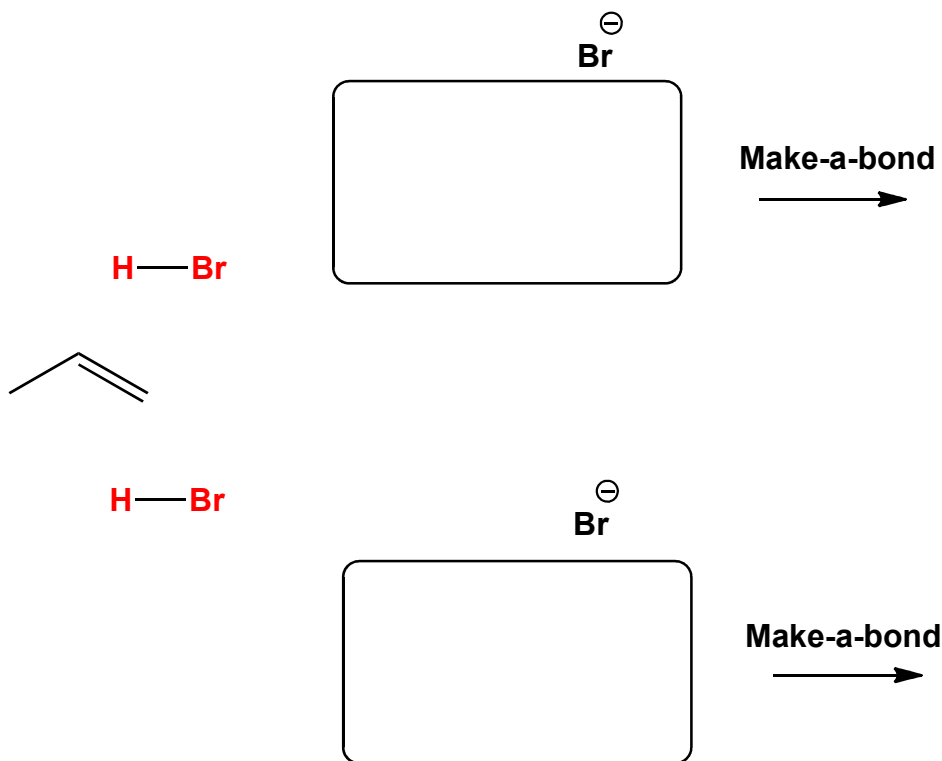


Reaction:

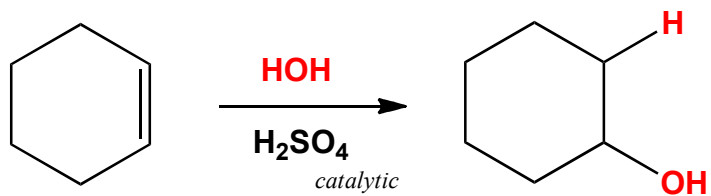


Regiochemistry: *Markovnikov's Rule*

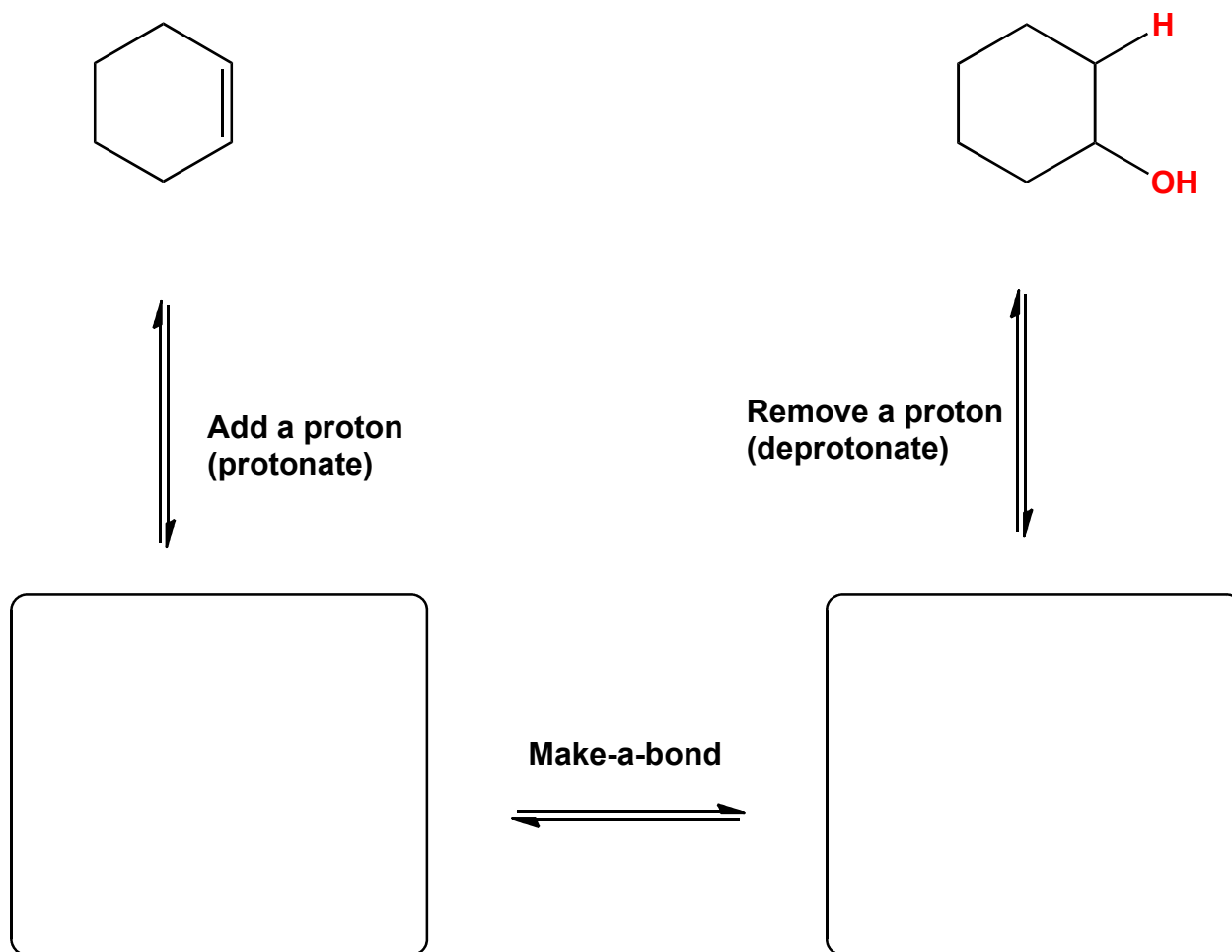
1) Add a proton--but where?



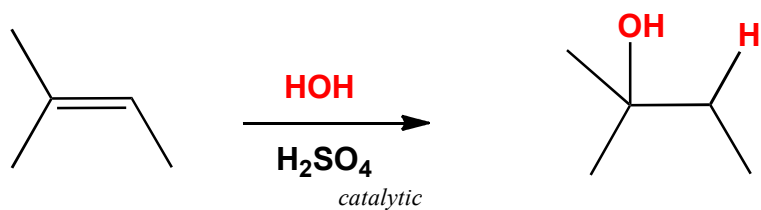
Reaction:



First-consider the acid/water reaction:



Reaction:



Regiochemistry: *Markovnikov's Rule*



Add a proton
(protonate)

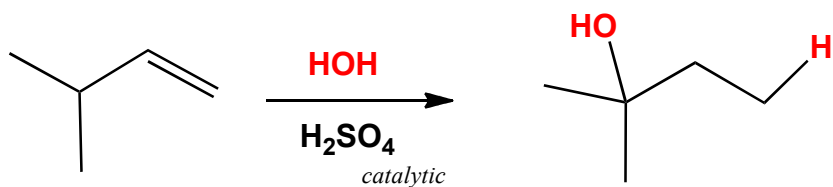


Remove a proton
(deprotonate)

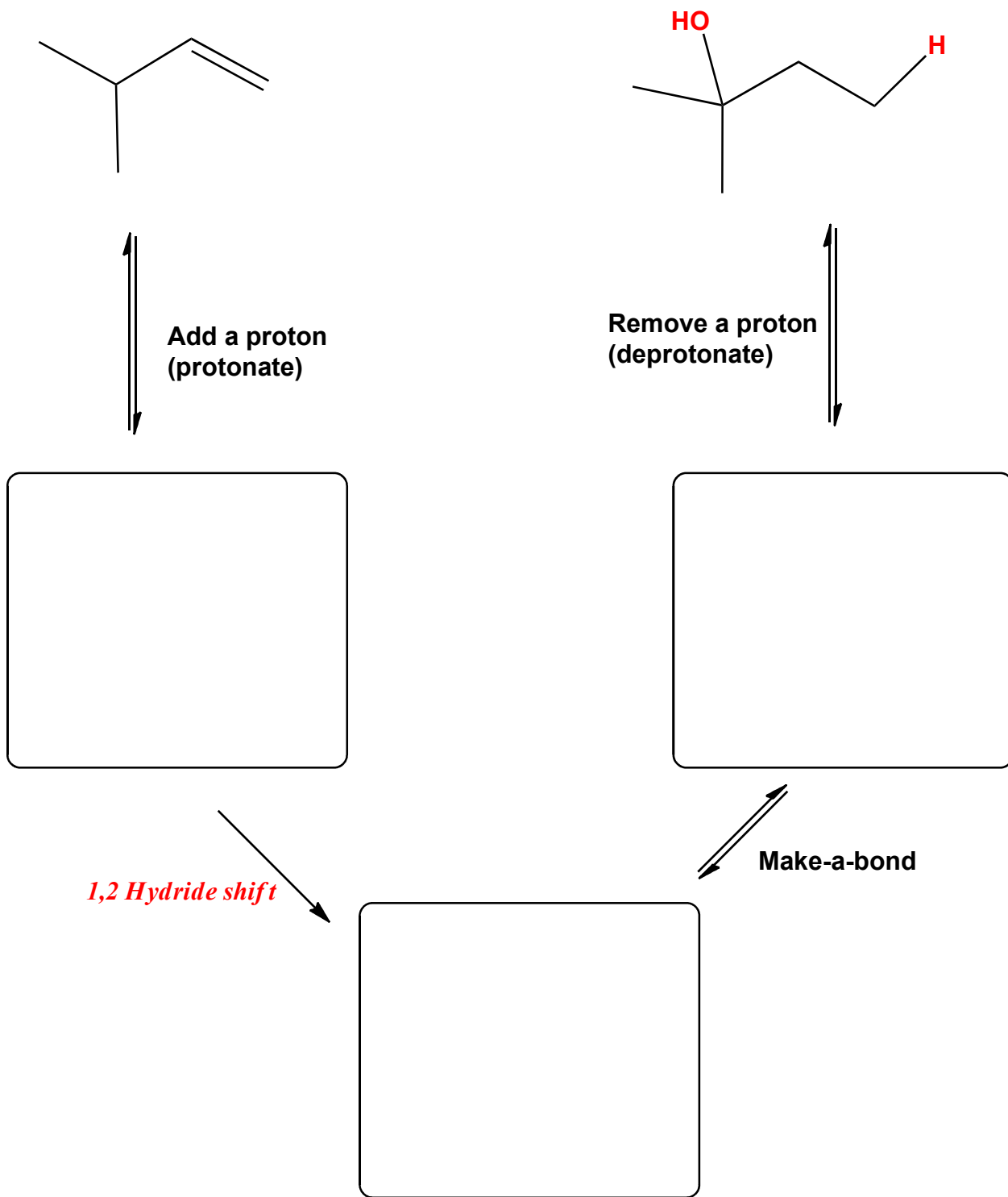


Make-a-bond

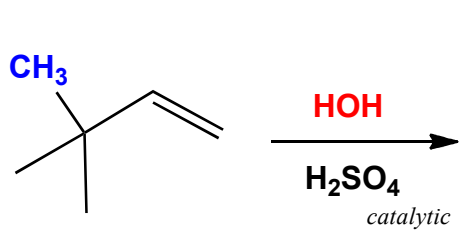
Reaction:



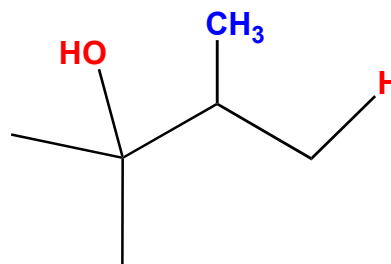
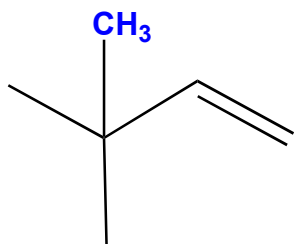
Note: rearrangement



Reaction:



Note: rearrangement



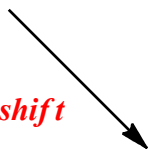
Add a proton
(protonate)



Remove a proton
(deprotonate)

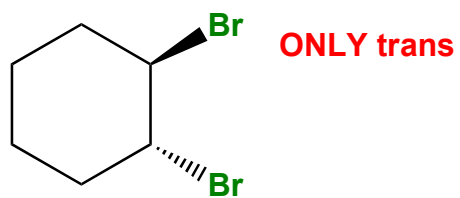
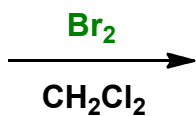
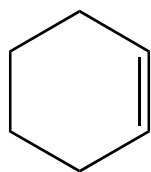


1,2 Methyl shift

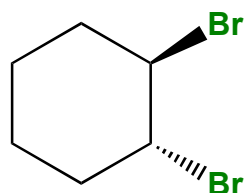
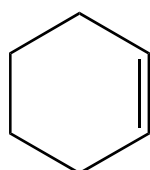


Make-a-bond

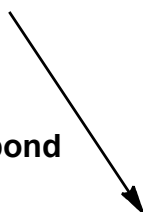
Reaction:



Stereochemistry: trans addition



Make-a-bond



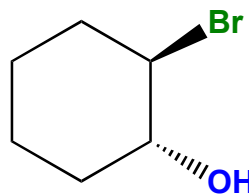
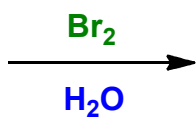
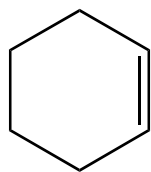
Make-a-bond



Bromonium Ion

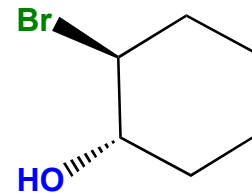
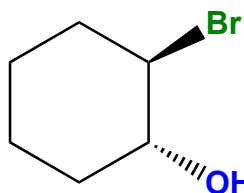
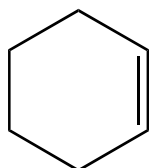


Reaction:



ONLY trans

Stereochemistry: trans addition

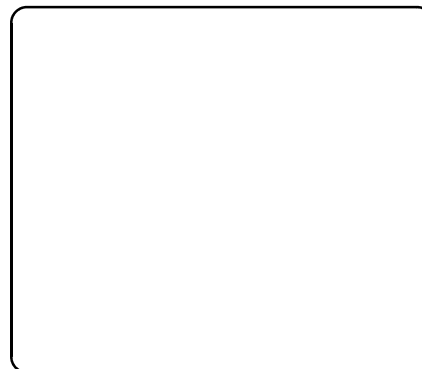


Make-a-bond



Bromonium Ion

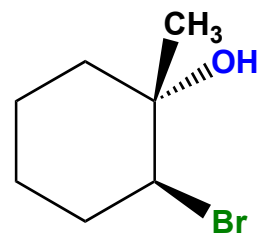
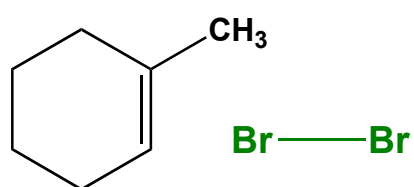
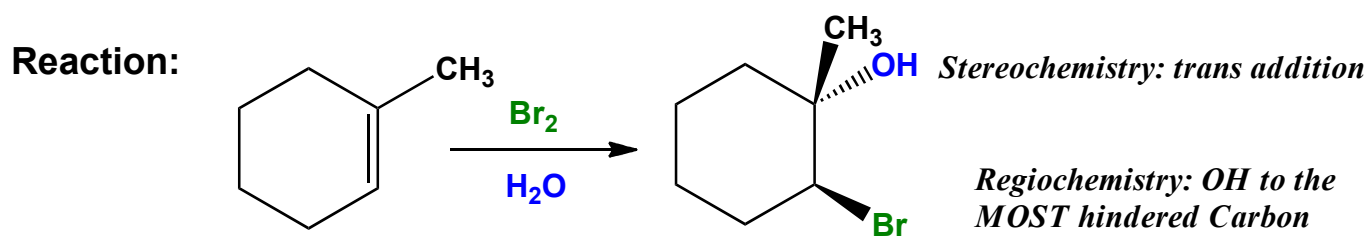
Make-a-bond



Oxonium Ion

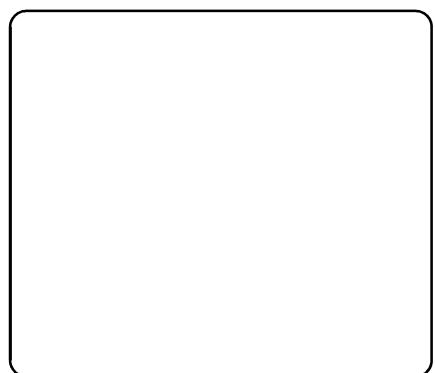
Remove a proton
(deprotonate)





Make-a-bond

Remove a proton
(deprotonate)

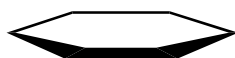
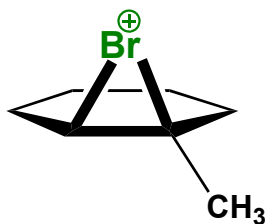


Make-a-bond

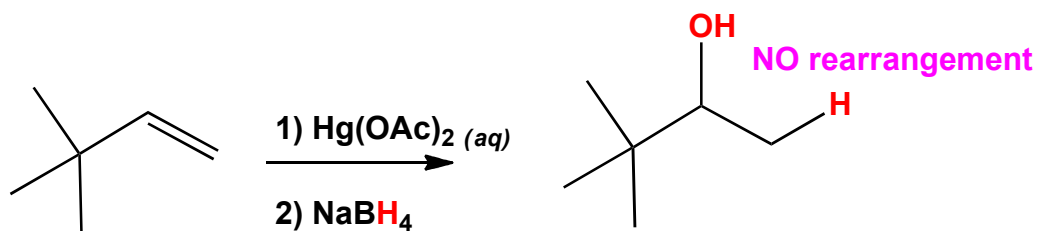


Bromonium Ion

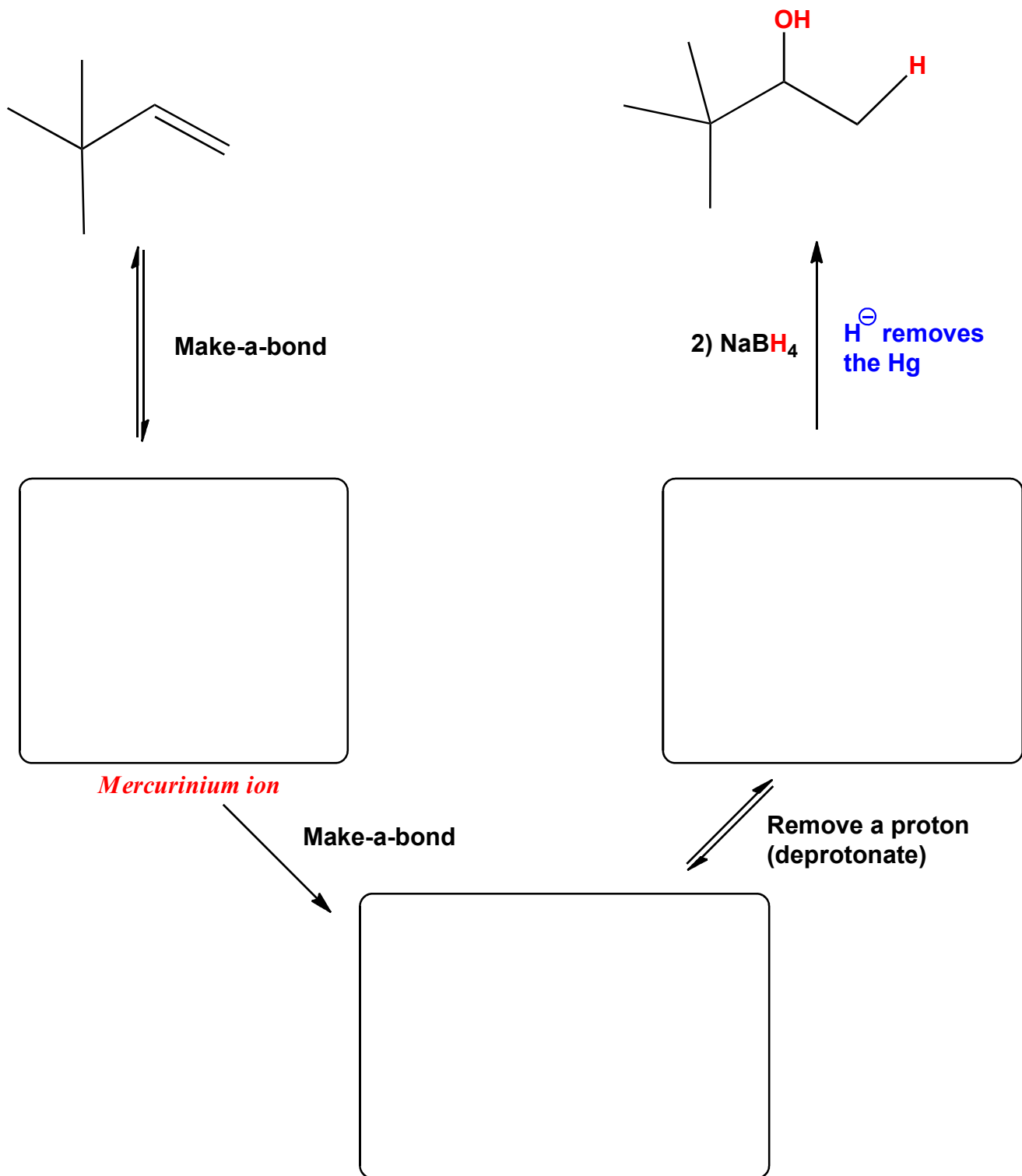
Oxonium Ion

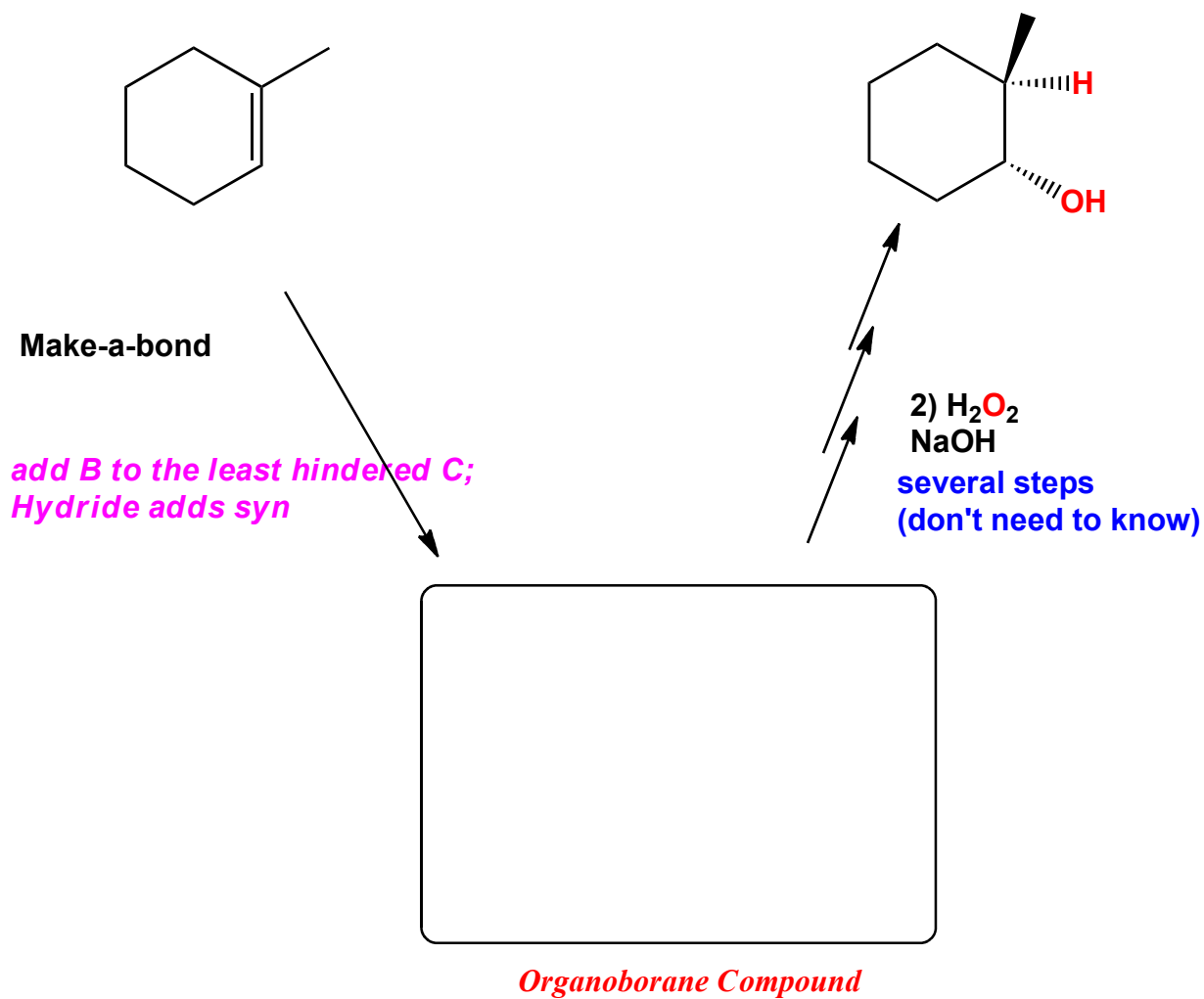
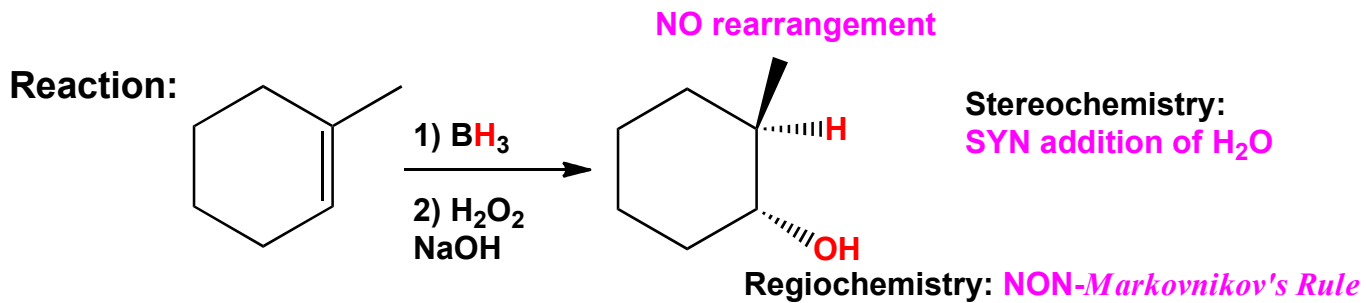


Reaction:

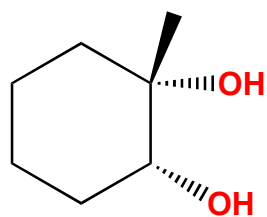
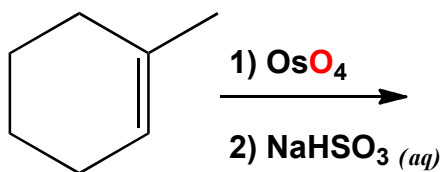


Regiochemistry: *Markovnikov's Rule*



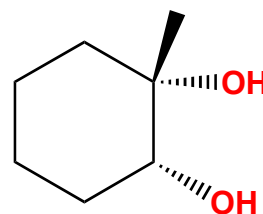
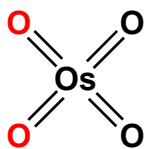
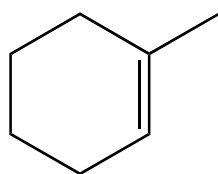


Reaction:



Glycols are the product

Stereochemistry:
SYN oxidation



Make-a-bonds

OsO₄ adds both O atom simultaneously

2) NaHSO_3

several steps
(don't need to know)



Osmate Ester

Sodium bisulfite

