| Last Name: | First Name: |
|------------|-------------|
| UTEID:     | Class Time: |

## Deadline for CH320/328 M HW 4: Oct 8 3:00 PM

1) Acid-Base reactions: Provide the products for each of the following reactions. Write out the mechanism for each reaction; use curved arrows to show the flow of pairs of electrons; show all formal charges.

$$H_3C \stackrel{\stackrel{}{\longrightarrow}}{\overset{}{\oplus}} H + OH \stackrel{\ominus}{\longrightarrow}$$

**2) Alkenes as "bases":** Draw the carbocation(s) from the protonation of each alkene. If there are two possible carbocations, indicate which one is more stable.





